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# Synthesis and X-ray crystallography of (1R,3aR,7aR)-1-((S)-1-((2R,5S)-5-(3-hydroxypentan-3-yl)tetrahydrofuran-2-yl)ethyl)-7a-methyloctahydro-4H-inden-4-one 

Modou Lo ${ }^{1}$, Andrea Martínez ${ }^{2}$, Hugo Santalla ${ }^{2}$, Fátima Garrido ${ }^{2}$, Aliou Hamady Barry ${ }^{3}$ and Mohamed Gaye ${ }^{1, *}$<br>${ }^{1}$ Department of Chemistry, University Cheikh Anta Diop, Dakar, 10700, Senegal<br>${ }^{2}$ Dipartamento Química Orgínica, Facultade de Química, Universidade de Vigo, Vigo, 36310, Spain<br>${ }^{3}$ Department of Chemistry, University of Nouakchott, Nouakchott, 130301, Mauritania<br>* Corresponding author at: Department of Chemistry, University Cheikh Anta Diop, Dakar, 10700, Senegal.<br>Tel.: +221.77.5555891. Fax: +221.33.8246318. E-mail address: mohamedl.gaye@ucad.edu.sn (M. Gaye).

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#### Abstract

The crystal of the title compound, $\mathrm{C}_{21} \mathrm{H}_{36} \mathrm{O}_{3}$ contains an oxolane ring, and six defined stereocenters which are unambigously established by the crystallography study. A three dimensional supramolecular architecture is ensured by hydrogen bonds from the hydroxy group which is both engaged in inter ( $\mathrm{O}-\mathrm{H} \cdots \mathrm{O}$ ) and intramolecular $\mathrm{C}-\mathrm{H} \cdots \mathrm{O}-\mathrm{H}$ ) hydrogen bonds. Weak $\mathrm{C}-\mathrm{H} \cdots \mathrm{O}=\mathrm{C}$ hydrogen bonds are involved also into the consolidation of the network.


## KEYWORDS

Calcitriol
Vitamin D
Crystal structure
NMR spectroscopy
Pyridinium dichromate
Heterocycle compounds

## 1. Introduction

Our constant interest in the chemistry of heterocyclic compounds and particularly in the synthesis of vitamin $D$ analogues, has led us to develop several methods for the synthesis of these compounds [1,2]. We also considered their biological activities which are studied in the literature [3]. Recently, we reported the synthesis of a new vitamin D2 analogue and the evaluation of its biological activity on colon cancer [4]. In the continuation of our work on the analogues of vitamin D, we synthesized two new molecules of calcitriol from an oxolane ring and its side chains [5]. In this study, we present the structure of a new analog of calcitriol with six stereo centers. The crystal structure allowed elucidating the absolute configuration of the stereo centers.

## 2. Experimental

### 2.1. Materials and physical methods

Diol and pyridinium dichromate (PDC) were purchased from Aldrich and used without further purification. The IR spectrum was recorded as KBr discs on a Bruker IFS-66 V spectrophotometer (4000-400 $\mathrm{cm}^{-1}$ ). Mass spectrometry was carried out with a Hewlett Packard 5988A spectrometer. The ${ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR spectra of the compound $\mathbf{1}$ were recorded in $\mathrm{CDCl}_{3}$ on a BRUKER 500 MHz spectrometer at room temperature using TMS as internal reference.

### 2.2. Crystal structure determination

Crystallographic data were collected at room temperature using a Bruker Smart 6000 CCD detector and $\mathrm{Cu}-\mathrm{K} \alpha$ radiation ( $\lambda=1.54178 \AA$ ) generated by a Incoatec microfocus source equipped with Incoatec Quazar MX optics. The software APEX3 [6] was used for collecting frames of data, indexing reflections and the determination of lattice parameters, SAINT [6] for integration of intensity of reflections, and SADABS [6] for scaling and empirical absorption correction. The structure was solved by dual-space methods using the program SHELXT [7].


Scheme 1


Figure 1. Crystal structure of the compound 1.

All non-hydrogen atoms were refined with anisotropic thermal parameters by full-matrix least-squares calculations on $F^{2}$ using the program SHELXL [8]. Hydrogen atoms were inserted at calculated positions and constrained with isotropic thermal parameters except for the hydrogen atom of the hydroxyl group. Drawings were produced with PLATON [9].

### 2.3. Synthesis of compound 1

To a solution of diol (2) ( 0.18 mmol ) in $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ ( 5 mL ), pyridinium dichromate (PDC) ( 0.37 mmol ) was added and the mixture stirred at room temperature for 12 h , then the solvent was evaporated and the residue was chromatographed on sılica gel using ( $10 \%$, EtOAc:hexane, $v: v$ ) to afford ketone (1) (Scheme 1). The title compound was recrystallized using a mixture of hexane:ethyl ether ( $1: 1, v: v$ ).
(1R, 3aR, 7aR)-1-((S)-1-((2R, 5S)-5-(3-hydroxypentan-3-yl)tetrahydrofuran-2-yl)ethyl)-7a-methyloctahydro-4H-inden-4-one (1): Color: White solid. M.p: $80-82{ }^{\circ} \mathrm{C}$. Yield: $88 \%$. Rf. 0.42 (30\%, EtOAc:hexane, $v: v$ ). $[\alpha]_{\mathrm{D}}{ }^{19}=+26.27\left(c 1.0, \mathrm{CDCl}_{3}\right) .{ }^{1} \mathrm{H}$ NMR ( $250 \mathrm{MHz}, \mathrm{CDCl}_{3}, \delta, \mathrm{ppm}$ ): $3.99(1 \mathrm{H}, \mathrm{td}, J=10.1,5 \mathrm{~Hz}, \mathrm{H}-$ $5^{\prime}$ ), $3.81(1 \mathrm{H}, \mathrm{dd}, J=9.8,5.7 \mathrm{~Hz}, \mathrm{H}-2 '), 2.42(1 \mathrm{H}, \mathrm{dd}, J=11.2,7.4$ $\mathrm{Hz}, \mathrm{H}-3 \mathrm{a}), 2.37-1.70\left(12 \mathrm{H}, \mathrm{m}, 6 \mathrm{x} \mathrm{CH}_{2}\right), 1.69-1.24(9 \mathrm{H}, \mathrm{m}, 3 \mathrm{x}$ $\left.\mathrm{CH}_{2}+\mathrm{H}-1+\mathrm{H}-1^{\prime}+\mathrm{HO}\right), 0.93\left(3 \mathrm{H}, \mathrm{d}, J=6.7 \mathrm{~Hz}, \mathrm{CH}_{3}-21\right), 0.87$ ( $6 \mathrm{H}, \mathrm{td}, J=7.5,5.3 \mathrm{~Hz}, \mathrm{CH}_{3}$-Et ), 0.66 ( $3 \mathrm{H}, \mathrm{s}, \mathrm{CH}_{3}-18$ ). ${ }^{13} \mathrm{C}$ NMR ( $62.5 \mathrm{MHz}, \mathrm{CDCl}_{3}, \delta, \mathrm{ppm}$ ): 211.91 (C=0), 83.49 (CH-2'), 81.78 (CH-5'), 74.76 (C-3'), 61.43 (CH-14), 54.23 (CH-17), 50.30 ( $\mathrm{CH}-13$ ), $41.02\left(\mathrm{CH}_{2}\right), 38.94\left(\mathrm{CH}_{2}\right), 38.93(\mathrm{CH}-20), 28.79\left(\mathrm{CH}_{2}\right)$, $26.91\left(\mathrm{CH}_{2}\right), 26.35\left(\mathrm{CH}_{2}\right), 25.92\left(\mathrm{CH}_{2}\right), 25.89\left(\mathrm{CH}_{2}\right), 24.05$ $\left(\mathrm{CH}_{2}\right), 19.26\left(\mathrm{CH}_{2}\right), 12.69\left(\mathrm{CH}_{3}-21\right), 12.50\left(\mathrm{CH}_{3}-18\right), 7.99\left(\mathrm{CH}_{3}-\right.$ Et), $7.60\left(\mathrm{CH}_{3}-\mathrm{Et}\right)$. IR ( $\mathrm{NaCl}, v, \mathrm{~cm}^{-1}$ ): 3491, 3361, 2964, 2879, 2347, 1713, 1460, 1381, 1245, 1145, 958, 890, 755. MS (ESI ${ }^{+}$) ( $\mathrm{m} / \mathrm{z}$, (\%)): $359.25\left(\left(\mathrm{M}^{+} \mathrm{Na}\right)^{+}, 47\right), 319.26\left((\mathrm{M}-\mathrm{OH})^{+}, 100\right)$. HRMS (ESI+): Calculated for $\mathrm{C}_{21} \mathrm{H}_{36} \mathrm{NaO}_{3}, 359.25567 \mathrm{~g} / \mathrm{mol}$; Found: $359.25562 \mathrm{~g} / \mathrm{mol}$.

## 3. Results and discussion

The compound 1 was prepared by a facile oxidation of compound 2 with pyridinium dichromate in dichloromethane (Scheme 1). Suitable X-ray crystals diffraction was obtained after recrystallization of compound $\mathbf{1}$ in a mixture of
hexane:ethyl ether ( $1: 1, v: v$ ). The afforded compound is soluble in common organic solvent such as chloroform. The mass spectrum of the compound 1 present a peak at 359.25562 amu corresponding to the molecular ion of $[1+\mathrm{Na}]^{+}$. The infrared spectrum of the compound shows absorption band pointed at $1713 \mathrm{~cm}^{-1}$ which is assigned to the $v(\mathrm{C}=0)$ vibration confirming the oxidation of the secondary alcohol function of compound 2 . In addition the ${ }^{13} \mathrm{C}$ NMR spectrum recorded in deuterated chloroform shows a characteristic signal at $\delta 211.91 \mathrm{ppm}$ which is assigned to the $\mathrm{C}=0$.

The molecular structure of the title compound is shown in Figure 1. Crystallographic data, selected bond lengths and angles, hydrogen-bond geometry and atomic displacement parameters are listed respectively in Table 1-4. The compound crystallizes in the non-centrosymmetric space group P2 $1_{1}$ and the absolute structure was unambiguously established. The molecule contains a cyclopentane ring trans-fused to a cyclohexanone ring. The lateral chain contains an oxolane ring. The cyclohexanone ring adopts a chair conformation. The cyclopentane ring is an envelope (Flap atom $=C 5$ ) and the tetrahydrofuran ring is twisted about C13-02. The configuretions of the stereogenic centres are C5 $(R), \mathrm{C} 6(R), \mathrm{C} 9(R)$, $\mathrm{C} 11(S), \mathrm{C} 13(R)$ and $\mathrm{C} 16(R)$. There is an intramolecular 0-H $\cdots 0$ hydrogen bond involving the hydroxyl group ( $03-\mathrm{H} 3$ ) and an oxolane 0 atom (02), generating an $S(5)$ ring motif (Figure 1 and Table 4). The bond lengths and angles are normal and comparable to those observed in compounds containing the bicyclic moiety fragment (1S,3aR,7aR)-1-ethyl-7a-methyl-octahydroinden-4-one); specially in our recent work [10], which concerns an isomer of the title compound, adopting a very similar crystal structure.

In the crystal, weak $\mathrm{C} 2-\mathrm{H} 2 \mathrm{~B} \cdots 01=\mathrm{C}$ hydrogen bonds (Table 4, Figure 2) link the molecules into C(4) chains, which propagate parallel to [101]. The chains are linked through extremely weak hydrogen bonds. Comparison of the crystal structure with that of the recent isomer we reported [10], shows that the two isomers adopt similar supramolecular architecture: hydrogen bonds linking molecules into chains, which propagate parallel to [101]. In both structures, the chains are linked by very weak H bonds.

Table 1. Crystal data and structure refinement for compound 1.

| Empirical formula |  |  | $\mathrm{C}_{21} \mathrm{H}_{36}$ |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: |
| Formula weight |  |  | 336.50 |  |  |
| Temperature ( K ) |  |  | 296(2) |  |  |
| Crystal shape / color |  |  | Block/ |  |  |
| Crystal system |  |  | Monoc |  |  |
| Space group |  |  | P2 ${ }_{1}$ |  |  |
| a (Å) |  |  | 12.415 |  |  |
| $\mathrm{b}(\AA)$ |  |  | 6.3672 |  |  |
| c ( $\AA$ ) |  |  | 12.718 |  |  |
| $\alpha\left({ }^{\circ}\right)$ |  |  | 90 |  |  |
| $\beta$ ( ${ }^{\circ}$ |  |  | 90.509 |  |  |
| $\gamma\left({ }^{\circ}\right)$ |  |  | 90 |  |  |
| Volume ( ${ }^{3}$ ) |  |  | 1005.4 |  |  |
| Z |  |  |  |  |  |
| $\rho_{\text {calc }}\left(\mathrm{g} / \mathrm{cm}^{3}\right)$ |  |  | 1.112 |  |  |
| $\mu\left(\mathrm{mm}^{-1}\right)$ |  |  | 0.562 |  |  |
| F(000) |  |  | 372.0 |  |  |
| Crystal size (mm³) |  |  | 0.111 | 0.053 |  |
| Radiation |  |  | CuK $\alpha$ | 78) |  |
| $2 \Theta$ range for data collection ( ${ }^{\circ}$ ) |  |  | 6.95 to |  |  |
| Index ranges |  |  | -15 $\leq h$ | $\leq k \leq 7$, |  |
| Reflections collected |  |  | 12352 |  |  |
| Independent reflections |  |  | 3777 | 296, $\mathrm{Rsi}_{\text {sig }}$ |  |
| Data/restraints/parameters |  |  | 3777/ |  |  |
| Goodness-of-fit on $\mathrm{F}^{2}$ |  |  | 1.065 |  |  |
| Final R indexes [ $1 \geq 2 \sigma$ ( I ] $]$ |  |  | $\mathrm{R}_{1}=0$. | ${ }_{2}=0.10$ |  |
| Final $R$ indexes [all data] Largest diff. peak/hole (e $\AA^{-3}$ ) |  |  | $\mathrm{R}_{1}=0.0$ | $2=0.10$ |  |
|  |  |  | 0.18/-0 |  |  |
| Table 2. Bond lengths for compound 1. |  |  |  |  |  |
| Atom | Atom | Length (Å) | Atom | Atom | Length (Å) |
| 01 | C1 | 1.213(3) | C7 | C8 | 1.542(3) |
| 02 | C16 | 1.437(2) | C8 | C9 | 1.551(3) |
| 02 | C13 | 1.447(3) | C9 | C11 | 1.538(3) |
| 03 | C17 | $1.431(3)$ | C11 | C12 | $1.522(4)$ |
| C1 | C6 | 1.501(3) | C11 | C13 | 1.540(3) |
| C1 | C2 | 1.504(3) | C13 | C14 | 1.528(3) |
| C2 | C3 | 1.519(4) | C14 | C15 | 1.525(3) |
| C3 | C4 | 1.531(3) | C15 | C16 | 1.519(3) |
| C4 | C5 | 1.532(3) | C16 | C17 | 1.529(3) |
| C5 | C10 | $1.539(3)$ | C17 | C20 | 1.528(3) |
| C5 | C9 | 1.551(3) | C17 | C18 | 1.531(3) |
| C5 | C6 | 1.553(3) | C18 | C19 | $1.506(4)$ |
| C6 | C7 | 1.514(3) | C20 | C21 | 1.522(4) |


| Atom | Atom | Atom | Angle ( ${ }^{\circ}$ ) | Atom | Atom | Atom | Angle ( ${ }^{\circ}$ ) |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| C16 | 02 | C13 | 109.87(16) | C5 | C9 | C8 | 103.55(16) |
| 01 | C1 | C6 | 123.5(2) | C12 | C11 | C9 | 114.17(19) |
| 01 | C1 | C2 | 123.2(2) | C12 | C11 | C13 | 111.22(18) |
| C6 | C1 | C2 | 113.3(2) | C9 | C11 | C13 | 109.33(17) |
| C1 | C2 | C3 | 113.63(19) | 02 | C13 | C14 | 105.57(17) |
| C2 | C3 | C4 | 113.8(2) | 02 | C13 | C11 | 110.18(18) |
| C3 | C4 | C5 | 111.88(19) | C14 | C13 | C11 | 116.21(18) |
| C4 | C5 | C10 | 110.61(19) | C15 | C14 | C13 | 102.53(19) |
| C4 | C5 | C9 | 117.21(17) | C16 | C15 | C14 | 101.03(18) |
| C10 | C5 | C9 | 111.04(17) | 02 | C16 | C15 | 104.17(17) |
| C4 | C5 | C6 | 106.79(17) | 02 | C16 | C17 | 107.87(16) |
| C10 | C5 | C6 | 111.02(18) | C15 | C16 | C17 | 118.20(19) |
| C9 | C5 | C6 | 99.50(15) | 03 | C17 | C20 | 106.00(18) |
| C1 | C6 | C7 | 119.95(19) | 03 | C17 | C16 | 109.04(17) |
| C1 | C6 | C5 | 111.81(17) | C20 | C17 | C16 | 109.75(19) |
| C7 | C6 | C5 | 105.13(17) | 03 | C17 | C18 | 108.45(18) |
| C6 | C7 | C8 | 103.97(17) | C20 | C17 | C18 | 112.73(19) |
| C7 | C8 | C9 | 107.11(16) | C16 | C17 | C18 | 110.69(18) |
| C11 | C9 | C5 | 119.18(17) | C19 | C18 | C17 | 116.4(2) |
| C11 | C9 | C8 | 112.71(17) | C21 | C20 | C17 | 114.7(2) |

The title isomer is characterized by the presence of an intramolecular hydrogen bond and the molecules in the chains are linked by weak H bonds. On the contrary, in the previously reported isomer, there is no intramolecular H bond and the molecules in the chains are strongly linked. It should be mentioned that the crystal structure change between the two isomers affected mainly the unit cell lattice, but not the crystal system (monoclinic), nor the space group symmetry ( $\mathrm{P} 2_{1}$ ). Indeed, the unit cell volume expands from the precedent isomer to the title isomer ( 979.3 to $1005.3 \AA^{3}$ ). This expansion resulted from contraction in the $c$ parameter and the increase
of the $a$ parameter, so that the $c / a$ ratio and the $\beta$ angle decreased from $1.8^{\circ}$ to $1.0^{\circ}$ and 104.2 to $90.5^{\circ}$, respectively. The $b$ parameter remains almost unchanged.

## 4. Conclusion

The titled compound having an oxolane moiety in his side chain was synthesized successfully and its structure has been determined by X-ray single crystallography. One of his stereoisomer was previously synthetized and reported by our group.

Table 4. Anisotropic displacement parameters $\left(\AA^{2} \times 10^{3}\right)$ for compound 1. The Anisotropic displacement factor exponent takes the form:

| Atom | $\mathrm{U}_{11}$ | $\mathbf{U}_{22}$ | $\mathbf{U 3 3}_{3}$ | $\mathrm{U}_{23}$ | $\mathrm{U}_{13}$ | $\mathrm{U}_{12}$ |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 01 | 84.8(11) | 73.9(11) | 58.6(9) | 5.9(8) | 7.4(8) | -0.2(9) |
| 02 | 79.6(10) | 49.9(8) | 55.5(8) | 10.5(7) | 15.0(7) | 10.9(7) |
| 03 | 54.1(8) | 99.9(14) | 69.2(9) | -1.4(10) | -7.4(7) | -4.6(9) |
| C1 | 64.2(12) | 50.8(12) | 60.3(12) | -0.1(10) | 10.6(10) | 2.5(9) |
| C2 | 55.6(12) | 72.6(15) | 80.1(15) | $2.6(13)$ | 14.0(11) | -3.9(11) |
| C3 | 51.9(12) | 77.9(16) | 90.0(17) | -3.0(14) | 3.0(11) | -13.8(12) |
| C4 | 53.3(11) | 69.7(14) | 65.3(12) | -1.8(11) | -4.6(9) | -7.1(11) |
| C5 | 47.8(10) | 46.6(10) | 53.4(10) | 1.5(9) | 0.0(8) | -1.7(9) |
| C6 | 51.2(10) | 51.5(11) | 55.2(11) | -0.5(9) | 2.6(8) | -1.2(9) |
| C7 | 55.7(12) | 91.1(18) | 54.4(11) | -8.7(12) | 0.9(9) | -8.4(12) |
| C8 | 49.1(10) | 72.7(14) | 58.0(11) | -1.3(11) | 1.9 (8) | -6.4(10) |
| C9 | 49.4(10) | 43.7(10) | 51.7(10) | 1.9 (8) | 0.3(8) | -1.2(8) |
| C10 | 75.8(15) | 50.5(12) | 72.4(14) | 6.3(10) | 10.3(12) | 2.8(11) |
| C11 | 61.9(12) | 45.8(10) | 54.2(10) | 1.6(9) | 6.5(9) | -1.7(9) |
| C12 | 87.3(17) | 81.1(18) | 61.6(13) | -13.9(13) | 5.7(12) | -24.3(14) |
| C13 | 58.9(11) | 53.9(12) | 52.3(10) | 2.6(10) | 5.4(9) | 1.4(9) |
| C14 | 68.5(13) | 47.4(11) | 65.8(13) | -4.0(9) | 6.2(11) | -2.6(10) |
| C15 | 74.0(14) | 47.5(11) | 62.8(12) | 4.5(10) | 4.6(10) | -6.5(10) |
| C16 | 55(1) | 45.3(10) | 55.5(10) | 7.5(8) | -1.9(8) | -1.2(8) |
| C17 | 50.7(10) | 52.7(11) | 53.2(10) | 5.7(9) | -2.6(8) | -4.3(9) |
| C18 | 63.8(12) | 52.2(11) | 63.2(12) | 6.4(10) | 8(1) | 2.4(10) |
| C19 | 81.3(17) | 70.5(16) | 92.6(19) | -22.7(15) | -2.2(14) | -9.7(14) |
| C20 | 82.4(16) | 56.4(13) | 64.6(13) | 10.5(11) | 5.7(12) | -12.0(12) |
| C21 | 116(2) | 91(2) | 71.2(16) | 7.6(16) | 25.6(16) | -26.4(19) |

Table 5. Hydrogen bonds for compound 1.

| $\mathbf{D}$ | $\mathbf{H}$ | $\mathbf{A}$ | $\mathbf{d ( D - H ) ( \AA )}$ | $\mathbf{d ( H - A ) ( \AA )}$ | $\mathbf{d ( \mathbf { A } ) \mathbf { A } ) ( \mathbf { A } )}$ |
| :--- | :--- | :--- | :--- | :--- | :--- |
| O3 | H3 | 02 | 0.82 | $2.766(2)$ |  |
| C2 | H2B | $011^{1}$ | 0.97 | $3.267(4)$ |  |
| $11-x,-1 / 2+y,-z$. |  |  | 2.55 | 112.6 | 130.7 |



Figure 2. Three dimensional network of compound 1.

In future, the biological activities of these two calcitriol analogues will be studied.

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## Supplementary material

Crystallographic data for the structure reported in this article have been deposited with Cambridge Crystallographic Data Center, CCDC-1555293. The data can be obtained free of charge at http://www.ccdc.cam.ac.uk/const/retrieving.html or from the Cambridge Crystallographic Data Centre (CCDC), 12 Union Road, Cambridge CB2 1EZ, UK; fax: +44(0)1223336033 or e-mail: deposit@ccdc.cam.ac.uk.

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